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Introduction and Motivation: Mineral grains on the surfaces of airless planetary bodies like the Moon are subjected to continuous radiation processing from solar energetic ions and also experience micrometeorite impact events. Together these processes alter the morphology, chemistry, and microstructure of surface particles and are collectively known as space weathering [1]. The optical properties of material on airless body surfaces are altered by space weathering, making it difficult to understand the chemical composition or mineralogy of that planetary body from remote sensing data [2]. Space-weathering features were first recognized in lunar soil samples returned by the Apollo missions e.g., [3]. These features primarily develop in the outer 100 nm of mineral grains exposed to interplanetary space, making transmission electron microscopy (TEM) uniquely suited for studying this surface process [3].

One of the primary characteristics of space weathering is the development of iron nanoparticles in the rims and interiors of mineral grains, typically <10 nm in size e.g., [3]. The oxidation state of iron in these particles, termed nanophase Fe (npFe⁰), can provide insight into the formation conditions of these features and the overall processes governing space weathering [4]. In this study we used electron energy-loss spectroscopy (EELS) coupled to an aberration-corrected and monochromated TEM to evaluate the oxidation state of individual npFe⁰ particles in lunar soils. Here we report results from six individual particles, which provide insight into the formation and processing history of grains on the surface of the Moon.

Samples and Methods: We embedded <1 mm grains of Apollo lunar soil sample 79221 in a low-viscosity epoxy and prepared samples for TEM analysis through ultramicrotomy. To complete our EELS analyses we used the 100 keV monochromated and aberration-corrected Nion UltraSTEM at Arizona State University equipped with a Gatan Enfinium spectrometer. We used a 3 mm spectrometer entrance aperture, a collection angle of 45 mrad and a 0.2 nm probe size (with a convergence semi-angle of 30 mrad). Our EELS spectra for individual particles were obtained from a line profile with a 0.1 eV/channel dispersion and an energy resolution between 200 to 300 meV. Each measurement had a variable dwell time, ranging between 0.01 s/px to 0.5 s/px. We collected EELS spectra for individual nanoparticles at the Fe₂,₃ core-loss edge. These spectra were compared to standards for both Fe⁰ and Fe²⁺ (expected oxidation states in these mineral samples) that had been previously acquired [4]. In order to qualitatively asses the oxidation state of npFe⁰, we can utilize data from previous experiments which suggest a ~1.5eV chemical shift between Fe⁰ and Fe²⁺.

Results and Discussion: Spectra collected from the particles and reference standards are shown in Figure 1. Comparison of the particle and reference spectra shows that the L₃ near-edge structure for the particles is intermediate between Fe⁰ and Fe²⁺ end members, indicating that they are composed of a mixture of oxidation states in varying proportions. Spectra a-c have primary peaks shifted to higher energies, more closely aligned with the Fe²⁺ standard, suggesting these particles contain more Fe²⁺. In
contrast, spectra e-g have peak positions shifted to lower energies, signifying a higher proportion of Fe$^0$ in these particles. An example of a particle from which we collected a spectrum is shown in Figure 2.

The EELS results suggest that current models for npFe$^0$ formation may not be complete as they predict that many of these particles condense from the vapor phase in reducing conditions. Our measurements show that some particles contain oxidized Fe, which suggests that the conditions may have been more oxidizing than previously thought. However, we note that the size of the nanoparticles (<10 nm on average) is smaller than the thickness of the ultramicrotomed slice (~60 nm), and so we cannot fully rule out at this time contributions to the spectrum from surrounding matrix material in which the nanoparticles are embedded. Understanding the contribution of the matrix material to the spectra will be important for determining, conclusively, the nature of the npFe$^0$ oxidation states.

References:

Figure 1: Fe L2,3 EELS spectra of six individual nanoparticles (a-c and e-g) and reference spectra for Fe$^{2+}$ (d) and Fe$^0$ (h) from [4]. Peak positions are indicated by vertical lines.

Figure 2: Left: High-angle annular dark field (HAADF) image of bright npFe$^0$ particles set in a relatively darker matrix. Red arrows indicate damage features caused by the probe. Right: HAADF image of a measured nanoparticle is identified by the red box. The dashed orange line indicates the location within the particle from which the spectrum was acquired. The dashed yellow line shows the outer rim of the grain.